

STUDIES ON ARUGOMYCIN, A NEW ANTHRACYCLINE ANTIBIOTIC
PART II. STRUCTURAL ELUCIDATION OF ARUGOMYCIN

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Summary: The total structure of arugomycin has been determined as shown in Fig. 1 based on spectroscopic data and chemical degradation.

In the previous paper¹⁾ we reported the structures of degradation products of arugomycin (AGM)²⁾, i.e. arugorol, AG1, AG2 and AG3 based on the ¹H- and ¹³C-NMR and mass spectral analysis. This paper describes structural determination of an additional hydrolysis product, AG4, and hydrogenolysis products, compounds A and B; these compounds gave enough experimental evidence to establish the total structure of AGM (Fig. 1).

The physicochemical properties of AGM are as follows; mp. 208-212°C, $[\alpha]_D^{25} +112^\circ$ (c 0.1, CHCl₃:MeOH = 9:1), C₈₀H₁₁₀O₃₇N₂, Anal. found: C 56.22, H 6.85, O 35.14, N 1.65 %, calcd.: C 56.71, H 6.67, O 34.97, N 1.65 %, SIMS m/z 1694 (MH⁺), IR ν_{\max}^{KBr} 3430, 2930, 1740, 1660, 1545, 1450, 1410, 1380, 1300, 1105 and 1000 cm⁻¹, UV $\lambda_{\max}^{\text{MeOH}}$ 235nm (ϵ 61500), 258(28300), 292 (10300) and 476 (17600).

In addition to the sp² carbons in the chromophore unit (Table 1, C-1 to C-13, -COOCH₃, C-1' to C-6' and (CH₃)₂N-), AGM contains two carbonyl (δ_C 166.0 and 168.9) and two olefinic methine carbons (δ_C 134.3 and 139.3) which were assigned to a fumaric acid residue based on comparison with the literature value³⁾. The olefinic protons observed as an AB-type quartet at δ_H 6.81 and 6.86 (J=17 Hz) are in agreement with this partial structure.

Methanolysis of AGM gave inter alia a methyl ester of methyl 4-O-fumaryl- α -L-diginoside⁴⁾, C₁₃H₂₀O₇. Since a resonance corresponding to the methyl ester signal (δ_H 3.81) in this compound was not found in the ¹H-NMR spectrum of AGM, and since AGM was methylated by diazomethane to give a methyl ester of AGM Me-AGM⁵⁾, C₈₁H₁₁₄O₃₇N₂, one of the carboxyl functions of the fumaric acid residue must be free in the parent compound. Consequently the 4-O-fumaryl- α -L-diginose moiety must be glycosidically combined to the terminal of a sugar chain.

The SI-mass spectrum of AG4⁶⁾ (C₆₂H₈₆O₂₈N₂) obtained as a partial hydrolysis product of AGM¹⁾ gave the protonated molecule at 1308 (MH⁺) which is

Table 1. 100MHz ^{13}C -NMR data of compound A (in CDCl_3), AG4 (in CDCl_3), AGM (in d_5 -pyridine)

	A	AG4	AGM		AG4	AGM
1	147.7	147.6	147.9	S-4 C-1	100.7	102.1
2	139.4	139.4	140.1	C-2	30.1	31.9*
3	121.8	121.7	120.1*	C-3	74.1	74.9*
4	155.9	155.5	156.6	C-4	75.3	75.9*
5a	116.6	116.4	117.8	C-5	67.9*	68.5*
6a	190.9	190.0	191.8	C-6	17.6*	17.6*
7a	112.6	113.2	114.8	-OCH ₃	56.8	56.1
8	159.8	161.4	162.3	S-5 C-1	99.2	100.0
9	231.0	71.2	70.6	C-2	32.4	34.4*
10	33.4	41.0	39.9	C-3	65.8	65.8
10a	70.5	70.7	69.3	C-4	67.9*	81.1
11	55.8	55.6	59.0	C-5	67.3**	67.4
11a	141.8	142.5	144.4	C-6	17.6*	17.2*
12	119.3	119.8	119.9	S-6 C-1		100.4*
13	114.6	114.8	115.3	C-2		31.9*
COO-	28.9	29.3	30.0	C-3		74.9*
-OCH ₃	173.0	171.7	171.9	C-4		75.3*
1	52.9	52.7	52.0	C-5		68.3*
2	97.4	97.3	97.9	C-6		17.6*
3	68.8	68.3	67.4	-OCH ₃		56.1*
4	80.1	81.3	82.7	S-7 C-1		100.0*
5	81.1	81.3	81.6	C-2		31.7*
6	77.2	77.5	77.3	C-3		73.8
NCH ₃	24.0	24.3	23.6	C-4		72.3
C=C	44.7	44.3	44.3	C-5		65.4
COOH			139.3	C-6		17.6*
COO-			168.8	-OCH ₃		55.9*
S-1 C-1	99.6	99.4	100.0			
C-2	33.9	33.7	34.6			
C-3	64.8	64.6	65.0			
C-4	82.9	82.5*	82.3			
C-5	67.5	67.8*	67.9			
C-6	17.0	16.8*	17.1			
S-2 C-1	99.1	99.2	99.7			
C-2	42.1	41.6	41.9			
C-3	88.7	88.5	89.5			
C-4	83.9	83.6	83.6			
C-5	69.0	70.3	70.6			
-CH ₃	19.0	18.8	18.6			
3	25.4	25.2	24.9			
S-3 C-1	101.6	101.4	102.3			
C-2	30.3	30.9	30.9*			
C-3	74.2	74.4	75.4*			
C-4	67.2	67.3*	67.5			
C-5	66.9	66.1*	67.2			
C-6	17.3	17.0	17.5			
-OCH ₃	55.8	56.3	56.1*			

*Tentative assignments

**Obscured by the solvent peak

larger than AG2¹⁾ by 130u. Since hydrolysis of AG4 gave 2-deoxyfucose(deFUC), diginose(DIG), and decilonitrose (DEC), this difference corresponds to the increment of a deFUC unit. The ^1H - and ^{13}C -NMR spectra of AG4 revealed five anomeric signals (δ_{C} 99.2, 99.2, 99.4, 100.7 and 101.4, δ_{H} 4.98, 5.01, 5.25, 5.32 and 5.52), two methoxy signals due to DIG (δ_{C} 56.2 and 56.3) and one characteristic tertiary CH_3 signal due to DEC (δ_{C} 25.2, δ_{H} 1.75 singlet). These data indicate that AG4 is a mono deoxyfucosyl derivative of AG2 comprising arugorol (ARG), DEC, deFUC and DIG in the ratio of 1:1:2:2. Comparison of the ^{13}C -NMR spectra of AG2 and AG4 revealed the glycosidation shift⁷⁾ of C-4 of DIG from δ_{C} 68.2¹⁾ to 75.3 in AG4. However, since there exist two terminal DIG units (S-3 and S-4) with almost identical ^{13}C chemical shifts in AG2, it was impossible to determine which of the terminal sugars in AG2 is glycosydated by deFUC(S-5) in AG4. Thus, AGM was subjected to catalytic hydrogenation (5 % Pd/BaSO₄) which resulted in the formation of hydrogenolysis products, compound A and compound B.

The ^{13}C -NMR spectrum of A⁸⁾ ($\text{C}_{49}\text{H}_{64}\text{O}_{21}\text{N}_2$) in CDCl_3 indicated the presence of a new methylene group at δ_{C} 20.7 due to C-7 of the chromophore and three anomeric carbons (δ_{C} 99.1, 99.6 and 101.6, δ_{H} 4.97, 5.25 and 5.33). The SI-mass spectrum of A showed the molecular ion peak at m/z 1018. These spectral data

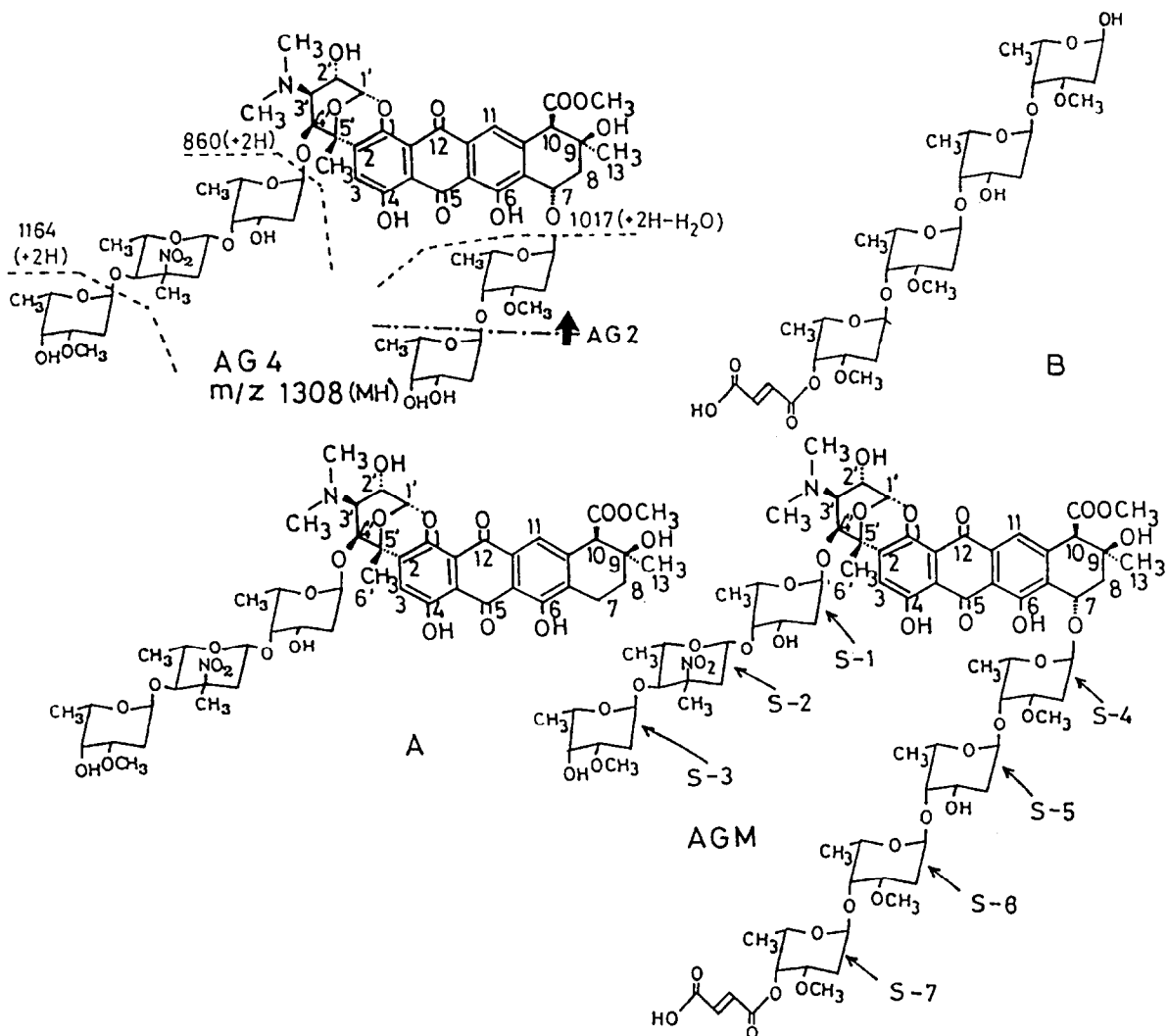


Fig. 1. SIMS-diagnostic ions of AG4 and the structures of AG4, A, B, and AGM.

proved the structure of A to be 4'-(L-diginosyl-L-decilonitrosyl-2-deoxyfucosyl)-7-deoxyarugorol (Fig. 1). Therefore, the additional deFUC(S-5) moiety is combined to C-4 of DIG(S-4) in AG2 and consequently, the structure of AG4 is illustrated as shown in Fig. 1. This conclusion was supported by detailed analysis of the SI-mass diagnostic ions at m/z 860, 1017, 1164 and 1308 (see Fig. 1). The anomeric configuration of deFuc(S-5) was determined to be α [H-1 of DIG(S-5); $J_{1,2ax}=2.0$, $J_{1,2eq}<1\text{Hz}$].

Compound B⁹⁾ (C₃₁H₅₀O₁₆) which was isolated as an anomeric mixture, gave DIG and deFUC on acid hydrolysis (40% HCOOH, 85°C, 40 min). The ¹H-NMR spectrum of B proved the presence of four sugars [δ_{H} 4.76, dd, $J=10.4$, 2.0 Hz (β -form) δ_{H} 5.42, dd, $J=2.0$, <1 Hz (α -form), δ_{H} 4.97, dd, $J=2.0$, <1 Hz, δ_{H} 4.99, dd, $J=2.0$, <1 Hz, δ_{H} 5.05, dd, $J=2.0$, <1 Hz] and three methoxy groups (δ_{H} 3.36, 3.41, 3.42) due to DIG. The IR spectrum of B gave absorption at 2400 - 2800, and 1720 cm⁻¹. Furthermore, B showed a characteristic AB-type quartet at δ_{H}

6.80 and 6.90 ($J=17$ Hz) arising from a fumaric acid half ester residue. Thus B consists of one mole each of fumaric acid and deFUC and three moles of DIG. Since the hydroxy group at C-3 of deFUC(S-5) in B was revealed to be free by $^1\text{H-NMR}$ spectral analysis of its diacetyl derivative (*vide infra*) and, since the linkage of deFUC to DIG has been established in AG4, the 4-O-fumaryl-DIG-DIG unit must be attached to C-4 of deFUC(S-5).

Reaction of B with acetic anhydride in pyridine gave an anomeric mixture of diacetate. Since the separation of these anomers was very difficult, the diacetate was analyzed without further purification. In its $^1\text{H-NMR}$ spectrum, two acylated proton signals were observed at δ_{H} 6.24 and 5.67 (the total area of these anomeric proton signals being equal to 1H), and 5.21. Irradiation of the proton at 5.21 collapsed the methylene signal (ddd) at δ_{H} 1.95 and 2.10 to double doublets* and thus, the acylated proton was unambiguously assigned to H-3 of 2-deFuc. Therefore, the fumaryl-Dig(S-7)-DIG(S-6) unit must be attached to deFUC(S-5) through an α -1,4 linkage and the structure of B is represented as shown in Fig. 1.

Based on these experimental results, the structure of AGM has been established as shown in Fig. 1 with some ambiguity about the absolute configuration.

* Due to the overlapping of complicated signals in this region, the changes of signals were detected by difference spectral method.

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REFERENCES AND FOOTNOTES

- 1) H. Kawai et al., The preceding paper.
- 2) H. Kawai et al., *J. Antibiotics*, **36**, 1569 (1983).
- 3) H. Seto et al., *Tetrahedron Lett.* **23**, 2667 (1982).
- 4) Methyl 4-O-fumaryl- α -L-diginoside methyl ester : $[\alpha]_{\text{D}}^{21} = -149^\circ$ (c 0.6, MeOH), M.W. (EI-mass) m/z 288 (M^+), oily substance.
- 5) Me-AGM : $[\alpha]_{\text{D}}^{22} = +219^\circ$, m.p. 214-215°C, M.W. 1708, $\text{UV}\lambda_{\text{max}}^{\text{MeOH}}$ 234 nm (ϵ 53500), 257(25100), 289(16600), and 478(14200).
- 6) AG4 : $[\alpha]_{\text{D}}^{22} = +330^\circ$, m.p. 209-210°C, M.W. 1307, $\text{UV}\lambda_{\text{max}}^{\text{MeOH}}$ 235 nm (ϵ 52700), 257(24300), 290(9500), and 478(16100).
- 7) S. Seo et al., *J. Amer. Chem. Soc.* **100**, 3331 (1978).
- 8) Compound A : $[\alpha]_{\text{D}}^{21} = +393^\circ$ (c 0.1, MeOH), m.p. 165-168°C, M.W. 1017, $\text{UV}\lambda_{\text{max}}^{\text{MeOH}}$ 235 nm (ϵ 54900), 248(28500), 292(11900), and 478(16000).
- 9) Compound B : $[\alpha]_{\text{D}}^{21} = -93.9^\circ$ (c 0.1, MeOH), m.p. 115-118°C.

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